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Large deformation of food gels: Influencing factors, theories, models, and applications—A review

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1 **Abstract**

2 Gels possess remarkable properties, and they hold particular importance in food science. After
3 consumption, food gels undergo large deformation, which impacts the overall texture of the
4 food. This process is influenced by various factors, including temperature, pressure, and
5 presence of crosslinking agents. Comprehensive insights into the interplay among these factors
6 and gel texture, combined with the theoretical exploration of gel deformation, enable the
7 development of foods to meet consumer preferences. To bolster the development of food gels,
8 in this review, we summarize the factors affecting the large deformation of gels. Moreover, we
9 discuss various mathematical models established by food scientists to explore the large
10 deformation of food gels and explore applications thereof. We expect that these insights into
11 the large deformation of gels can lead to their increased utilization in the food industry.

12 **Keywords:** food gels; large deformation; theoretical basis; application

13 **1. Introduction**

14 Amidst global sustainability challenges, particularly those arising from population growth,
15 environmental pressures, and energy consumption, development of innovative food products
16 that address human nutritional requirements has become increasingly critical ([Kell, 2022](#);
17 [Kumar et al., 2022](#)). Among these innovations, gels have emerged as a promising solution. As
18 elastic, semisolid soft materials with a three-dimensional (3D) network structure, gels
19 demonstrate excellent controllability and large deformation characteristics ([Gul et al., 2022](#);
20 [Yuan et al., 2025](#)). The term “large deformation” indicates irreversible changes in the volume
21 and shape of a gel due to external stimuli such as temperature, pH, or mechanical load, resulting
22 in the fracture or collapse of the gel matrix and the permanent disruption of its molecular
23 network. This property not only determines the performance of gels during food processing but
24 also considerably affects the texture of food products ([Gheorghita Puscaselu et al., 2020](#)). Gels
25 are used in a wide range of foods, from desserts, yogurts, to coatings for baked goods and fruits
26 ([Silva et al., 2020](#); [Górska et al., 2024](#)). In addition, they can be used as fat substitutes to
27 maintain the nutritional value of foods and reduce their caloric content, thereby decreasing the
28 risk of chronic conditions such as cardiovascular diseases and obesity ([Nath et al., 2023](#);
29 [Nepovinnykh et al., 2019](#); [Yang et al., 2020](#)). However, the large deformation properties of gels,
30 which are essential for their structural and functional optimization, remain poorly understood.

31 In laboratory settings, research on gel deformation primarily focuses on texture and
32 rheological properties of gels. Texture analysis involves the assessment of various
33 characteristics, such as stretching, compression, and puncture ([St. Pierre et al., 2024](#)), while
34 rheological analysis primarily focuses on the viscoelastic properties and stress–strain responses
35 of gels ([Y. Wang & Selomulya, 2022](#)). Modern rheometers can precisely determine the manner
36 in which complex materials react to applied stress or strain, enabling the characterization of
37 gel composition, structure, and processing effects. Customized gels with required

38 characteristics can be fabricated by manipulating the polymer microstructure and surrounding
39 medium (Nath et al., 2023). The complete rheological deformation curve of a viscoelastic gel
40 typically comprises linear, nonlinear, and fracture regions (Fig. 1). Deformation is minimal in
41 the linear region, whereas the nonlinear region shows more substantial deformation and is thus
42 termed the “large deformation” region (Zhang et al., 2007). Under industrial processing
43 conditions, gels are exposed to high levels of applied stress and strain. In such cases, mimetic
44 mechanical tests prove to be invaluable for rheological characterization (Walayat et al., 2022).

45 In a previous review, Sinha et al. (2024) discussed a novel bi-gel system that combined
46 the properties of organogels and hydrogels, focusing on how adjusting the organogel and
47 hydrogel fractions can regulate their properties. Meanwhile, Lin et al. (2020) comprehensively
48 reviewed the preparation methods for different types of emulsion gels (including bulk emulsion
49 gels, emulsion gel particles, and fluid emulsion gels), their structure–property relationships,
50 and their applications in the food industry. Moreover, B. Liu et al. (2024) systematically
51 classified the mechanisms of food gel synthesis and discussed the application potential of gels
52 in various functional foods, highlighting their applications in food packaging, satiety gel
53 systems, nutrient delivery systems, food coloring adsorption, and food safety monitoring.
54 While considerable research has been conducted on gels in food science, there remains a critical
55 gap in understanding the manner in which the large deformation properties of these gels can be
56 optimized for diverse applications. Although the existing research predominantly focuses on
57 isolated factors such as pH, temperature, and crosslinking mechanisms, the interplay among
58 these factors remains underexplored. Moreover, advancements in theoretical modeling and
59 experimental tools have yet to be fully integrated into practical applications, limiting their
60 potential effect on food innovation. Therefore, this review examines the factors influencing the
61 large deformation of gels and discusses the mathematical models established by food scientists
62 to explore this deformation in the context of food gel performance. This review aims to

63 synthesize the current knowledge regarding the large deformation properties of gels, highlight
64 the research gaps, and propose future directions to bridge the divide between fundamental
65 research and industrial applications.

66 **2. Theoretical framework for the large deformation of gels**

67 The large deformation behavior of gels involves two key phenomena: (1) nonlinear elastic
68 response, in which the relationship between force and deformation becomes nonlinear, and (2)
69 material failure at a critical strain, referred to as the breaking strain. Both these conditions can
70 be studied using stress–strain curves (Bot et al., 1996). Damage and fracture both need to be
71 modeled to accurately simulate gel behavior. However, one key challenge lies in understanding
72 the effect of fluid content on gel fracture. In addition, many models underestimate fracture
73 resistance because they ignore the energy dissipation caused by fluid diffusion (Lake & Thomas,
74 1967). To address these issues, theoretical models and simulations need to incorporate fluid
75 diffusion alongside the large deformation, damage, and fracture mechanisms that characterize
76 gel behavior.

77 Mao and Anand (2018) developed a theoretical framework to explain the fracture of
78 polymeric gels, incorporating the combined effects of fluid diffusion, large deformation, and
79 damage. Their theory introduced two key ideas. First, it accounted for the changes in a gel's
80 free energy due to entropy and internal-energy shifts caused by the stretching of polymer chains.
81 Second, it posited that gel damage occurs through chain-scission, in which polymer chains
82 break because of internal-energy changes rather than configurational entropy changes.

83 Small molecules can migrate into polymer networks, causing the network to swell and
84 form polymeric gels. Hong et al. (2008) developed a theory combining mass transport and large
85 deformation to explain this behavior. This theory suggested that the free energy of the gel arises
86 from two molecular processes: the stretching of the polymer network and its mixing with small
87 molecules. Small molecules and polymer chains both were treated as incompressible, and

88 osmotic pressure was modeled using a Lagrange multiplier. The theory demonstrated that gels
89 can deform through (1) rapid local rearrangement of molecules, which changes their shape
90 without altering their volume, or (2) slower migration of small molecules, which affects both
91 shape and volume. These mechanisms were modeled assuming that small molecules diffuse
92 within the gel. This theory explains gel behavior under various conditions, such as those when
93 a constrained gel layer is subjected to weight or deformation by means of a conical indenter.
94 Such advancements enable more accurate modeling and a deeper understanding of gel
95 deformation in complex scenarios.

96 *2.1. Theoretical basis of the large deformation of gels*

97 *2.1.1. Free energy functions*

98 Theories for hydrogel deformation were first proposed in pioneering works by [Flory and](#)
99 [Rehner \(1943\)](#). These initial theories suggested that the Helmholtz free energy of a polymer
100 network in an aqueous solution can be represented as the additive decomposition of the free
101 energy of the elastic stretch of the polymer network and the free energy of mixing when the
102 polymer network interacts with a solution. This decomposition has been widely used in
103 subsequent theories of hydrogel behavior. [Yang and Wang \(2019\)](#) developed governing
104 equations for functional gradient spherical hydrogels under spherically symmetric conditions
105 by using the Flory-Huggins free energy function to simulate the nonuniform and large
106 deformation swelling behavior of these hydrogels at a given internal pressure and chemical
107 potential. According to the Flory theory, the more general form of the free energy of a hydrogel
108 can be described as follows:

$$109 \quad W = W_{net} + W_{mix} \sum_{r \neq s} W_r \quad (1)$$

110 where W_{net} and W_{mix} denote the free energy of network stretch and that of mixing, respectively.
111 These two entities are applicable to all types of hydrogels. Meanwhile, W_r represents the free
112 energy contributions of factors other than the solvent and is specific to certain types of gels.

113 These free energies include, but are not limited to, free energy of ionization (W_{ion}), polarization
114 (W_{pol}), and dissociation (W_{dis}), which have been detailed previously (Z. Liu et al., 2015) and
115 thus have not been repeated in this review.

116 2.1.2. Coupled diffusion–deformation theory

117 Baek and Srinivasa (2004) established thermodynamic theories of coupled diffusion and
118 large deformation, paving the way for future continuum mechanics studies of hydrogel swelling.
119 Yang et al. (2022) investigated the transient behavior of diffusion and deformation in hydrogels
120 on the basis of the coupled diffusion theory. They stated that the diffusion of solutes from a
121 spherical hydrogel occurs under the influence of fluctuating stresses. Meanwhile, the corners
122 of a square hydrogel swell faster than the edges and interior, leading to deformation. Recent
123 developments in coupled diffusion–deformation models have addressed some of these gaps by
124 integrating solvent movement with mechanical deformation. However, these models often
125 require extensive computational resources and are yet to be experimentally validated in
126 complex food systems. Moreover, most existing models assume uniform material properties,
127 limiting their accuracy in describing hybrid gels or multicomponent systems, both of which are
128 commonly used in food formulations.

129 The equations governing gel swelling depend on the migration of solvent molecules. The
130 flux is expressed (Bouklas & Huang, 2012) as follows:

$$131 \quad j_i = \frac{cD}{kT} \frac{\partial \mu}{\partial x_i} \quad (2)$$

132 Moreover, it follows conservation laws:

$$133 \quad \int_V \frac{1}{\det F} \frac{\partial C}{\partial t} dV + \int_A j_i dA = 0 \quad (3)$$

134 where K is the Boltzmann constant, T is the temperature, V is the reaction volume, A is the
135 reaction area, t is the time, C is the solvent concentration, D is the constant of solvent diffusivity,
136 F is the deformation gradient, μ is the chemical potential, and c is the true solvent concentration.

137 2.1.3. Theories for large deformation of hydrogels

138 Under the influence of external stimuli, a hydrogel undergoes deformation from its initial
 139 reference state to its final current state. In describing this process of deformation, the
 140 deformation gradient F_{iK} (or F) is conventionally used. Fig. 2 shows the state of deformation
 141 of a hydrogel. [Dutta et al. \(2020\)](#) examined the hygroscopicity of white rice using deformation
 142 gradient equation modeling to identify a set of optimal process parameters in a temperature
 143 range of 25°C–80°C.

144 In the current state at time t , the marker X moves to a place with coordinates $x(X,t)$. The
 145 deformation gradient is denoted ([Z. Liu et al., 2015](#)) as follows:

$$146 \quad F_{iK}(X, t) = \frac{\partial x_i(X,t)}{\partial X_K} \quad (4)$$

147 Under conditions of equilibrium, the change in the free energy of the hydrogel is balanced
 148 via the external work performed on the gel. This thermodynamic equilibrium is usually
 149 expressed ([Toh et al., 2014](#)) as follows:

$$150 \quad \int_V \delta W dV = \int_V B_i \delta x_i dV + \int_A T_i \delta x_i dA + \sum (\mu^r \int_V \delta C^r dV) \quad (5)$$

151 where W is the free energy of the gel; B_i is the external force; T_i is traction; and μ^r and C^r are
 152 the chemical potential and concentration of the r species, respectively.

153 2.2. Numerical simulation of hydrogel behavior

154 Numerical simulation is often employed to predict the behavior of hydrogels because it
 155 allows for us to study gels with complex geometries, which are often excessively complicated
 156 to study via analytical methods. The common simulation tools employed to study hydrogel
 157 behaviors include the finite element method, meshless methods, and molecular dynamics
 158 simulation.

159 2.2.1. Finite element method

160 The finite element method is a numerical solution method used for solving problems in
 161 elastic mechanics ([Y. Liu et al., 2021](#)). This method was rapidly developed following the
 162 availability of the electronic computer. Gels are soft materials and can undergo large

163 deformations, and their mechanical behavior is influenced both by their polymer and solvent
164 compositions. Currently, experimental methods are insufficient to accurately determine the
165 mechanical properties of gel materials. Therefore, the finite element method has emerged as an
166 important and commonly used simulation tool for study of gel deformation. With the
167 development of monophasic theories, the finite element simulation of hydrogel swelling has
168 been expedited even further. ABAQUS and COMSOL Multiphysics, two finite element
169 software, are currently the most popular tools employed for these simulations. ABAQUS can
170 define material models through user-defined subroutines, while COMSOL Multiphysics is
171 often used for the simulation of heavily coupled physical material models (Wu et al., 2023).

172 Nakauma et al. (2014) used a finite element model to simulate the compression of agar
173 gel on different substrates, using linear elastic properties for their analysis. Their simulation
174 results were validated through experimental findings. Meanwhile, Y. Liu et al. (2015)
175 developed a finite element algorithm to model the uneven swelling of neutral hydrogels at
176 equilibrium. In addition, Toh et al. (2014) employed the finite element method to simulate large,
177 uneven deformations in gels under geometric constraints. Interestingly, using a custom
178 subroutine in ABAQUS, they modeled gels as hyperplastic materials. These studies showed
179 that finite element models can accurately predict the deformation of photothermal gels with
180 complex shapes.

181 2.2.2. Ogden model

182 The Ogden model (Bergström, 2015) is a very general mathematical hyperelasticity model
183 in which the Helmholtz free energy per reference volume is expressed in terms of the applied
184 principal stretch. The Ogden model can be used to describe the mechanical behavior of
185 materials under conditions of large deformation. This model is particularly suitable for
186 hyperelastic materials, such as rubber and some polymers, as these materials exhibit nonlinear
187 stress–strain relationships under the influence of large strains.

188 The first-order Ogden model expresses the mathematical relationship of compression
189 stress with gel concentration, applied to fit the experimental data (Yao et al., 2022):

$$190 \quad \sigma = \frac{2\mu}{\alpha} \left(\lambda^\alpha - \lambda^{\left(-\frac{\alpha}{2}\right)} \right) \quad (6)$$

191 where σ is the shear stress, μ is the shear modulus, and α is the Ogden constant.

192 The Ogden constitutive model has been widely adopted to describe the hyperelastic
193 behavior of soft materials such as hydrogels and living tissues in the presence of uniaxial
194 compression (Nedrelov et al., 2023). The first-order Ogden model was observed to be well-
195 fitted to the effect of carbohydrates on the large deformation behavior of fish skin gelatin (X.
196 Li et al., 2020). Moreover, Czerner et al. (2016) used the Ogden model to study the deformation
197 and fracture behavior of physical gelatin gel systems and determined the shear modulus and
198 strain hardening capability of the gels by means of the stress–stretch ratio curve.

199 2.2.3. BST model

200 In 1974, Blatz et al. (1974) proposed a strain energy density function based on a
201 generalized measure of strain, which was later employed to study the large deformation of gels.
202 This framework is referred to as the BST model. The BST model was derived from the rubber
203 elasticity theory and has been employed to characterize the curvature of the stress–strain profile
204 beyond the linear range. This model has been successfully applied to describe the deformation
205 mechanics of various biopolymer food gels, including alginate (Zhang et al., 2005), mixed soy
206 and κ -carrageenan (Cavallieri et al., 2010), and gelatin gels (Gravelle & Marangoni, 2021):

$$207 \quad \sigma = \frac{2E}{3n} (\lambda^n - \lambda^{-2n}) \quad n \sim \frac{d_f}{d_f - 1} \quad (7)$$

208 where σ is the shear stress, λ is the stretch ratio, E is Young's modulus, and n is an elasticity
209 parameter.

210 Theoretical models such as the Ogden and BST frameworks have significantly advanced
211 the understanding of gel deformation, particularly in terms of comprehending nonlinear

212 elasticity and fracture behavior. However, their application in food systems remains limited
213 because of several challenges. The Ogden model, while effective at describing nonlinear stress–
214 strain relationships, does not account for time-dependent viscoelastic properties, which are
215 critical in dynamic processes such as extrusion and chewing. Similarly, the BST model offers
216 robust predictions of gel fracture but struggles to incorporate solvent migration and
217 environmental factors such as pH and temperature variations.

218 Thus, future advancements should focus on integrating these theoretical frameworks with
219 machine learning and experimental data so as to improve their predictive power and practical
220 relevance. For instance, machine learning could be employed to identify patterns in large
221 datasets generated in rheological experiments, enabling models to better comprehend the
222 multiscale behavior of gels. Moreover, hybrid models that combine microstructural dynamics
223 with macroscopic deformation behavior might provide a more comprehensive understanding
224 of gel mechanics under real-world conditions. By addressing these limitations, theoretical
225 models might play a pivotal role in optimizing gel properties for applications such as 3D food
226 printing, the development of dysphagia-friendly formulations, and bionic food design, bridging
227 the gap between fundamental research and industrial practice.

228 **3. Factors affecting the large deformation of gels**

229 Various factors can influence the large deformation behavior of gels and can be split into
230 two groups: internal and external. Notably, the internal factors are intrinsic to the gel and
231 include gel concentration and gel composition. Meanwhile, the external factors include pH,
232 temperature, high-pressure processing (HPP), crystal addition, crosslinking agents, and
233 polysaccharide charge density. Food texture is affected by the large deformation characteristics
234 of food gels ([Gravelle & Marangoni, 2021](#)). Understanding the manner in which these factors
235 influence food properties might aid the development of novel food products customized to
236 consumer demands.

237 3.1. Gel composition

238 Gel composition is an important intrinsic factor affecting the mechanical strength of gels.
239 Proteins and polysaccharides are the most common gel components used in food products and
240 have attracted considerable attention in the field of functional foods because of their low
241 toxicity, edibility, biocompatibility, biodegradability, and affordability (K. Liu et al., 2021).

242 Polysaccharides strengthen gels by enhancing hydrogen bonding and forming robust
243 double-network structures. For example, viscoelastic analysis has shown that curdlan expedites
244 the formation of denser rice starch gels with a higher degree of solidity (Wang et al., 2024).
245 Similarly, *Mesona chinensis* polysaccharide–tapioca starch gels exhibit excellent flexibility
246 and durability, with a stable network structure maintained via hydrogen bonding (Huang et al.,
247 2023).

248 Gelatin, a type of protein, facilitates the formation of hydrophobic interactions, disulfide
249 bonds, and hydrogen bonds among globular proteins, transforming weak 3D gel networks into
250 dense-layered structures. Gelatin can be categorized as aquatic or terrestrial depending on its
251 source. In particular, aquatic gelatin is primarily derived from fish (fish skin, fish scales, and
252 swim bladders), while terrestrial one is usually derived from the skin, bones, and connective
253 tissues of swine and cattle (Derkach et al., 2022; Nurilmala et al., 2021; Zou et al., 2024).
254 Among the different types of gelatin extracted from animal skins (pig skin, cow skin, and fish
255 skin), pig skin and cow skin gelatin exhibit the greatest and poorest ability to resist deformation,
256 respectively (Michelini et al., 2020). The gel strength of bovine bone gelatin, however, is
257 greater than that of fish skin gelatin (T. Zhang et al., 2020). These variations highlight the
258 manner in which the source and amino-acid composition of gelatin affect its ability to resist
259 large deformations. The energy storage modulus of globular protein gels significantly increases
260 when gelatin is added, suggesting that gelatin can improve the strength of these gels and
261 stabilize the protein network structure (Ma et al., 2022).

262 The choice of gel components—proteins or polysaccharides—greatly influences the
263 mechanical and sensory properties of a gel. While polysaccharides facilitate robust network
264 formation via hydrogen bonding, proteins provide a tunable rheological behavior. Previous
265 research has largely focused on these components in isolation, neglecting the manner in which
266 hybrid gels comprising proteins and polysaccharides may offer superior large deformation
267 characteristics. Thus, comparative studies on hybrid gel systems could facilitate the
268 development of novel gel formulations for customized applications.

269 *3.2. Gel concentration*

270 Gel concentration is another intrinsic factor affecting the mechanical strength of gels. The
271 minimum protein concentration required to form gels varies from one protein to another. For
272 example, the minimal concentration required for gel formation is 0.6% for gelatin, 3% for egg
273 albumin, 6.6% for soy proteins, and 4%–12% for whey proteins and varies depending on the
274 pH and ionic strength (Le et al., 2017). In particular, gel concentration affects gel strength by
275 altering the gel's microstructure. In low-concentration gels, the microstructure is loose and the
276 pore size is large. Meanwhile, high-concentration gels are dense and have a small pore size
277 (Cortez-Trejo et al., 2021). Horinaka et al. (2022) studied the effect of gel concentration on the
278 cyclic deformation behavior of κ -carrageenan hydrogels. High-concentration κ -carrageenan
279 hydrogels exhibited good elastic properties, whereas low-concentration κ -carrageenan ones
280 showed a higher Young's modulus. Subsequently, Horinaka and Ogawa (2023) also explored
281 the effect of agarose concentration on the cyclic deformation behavior of agarose hydrogels,
282 and their findings were consistent with those related to κ -carrageenan hydrogels.

283 Gel concentration considerably impacts the mechanical properties and deformation
284 behavior of gels. However, while high-concentration gels often exhibit improved strength and
285 reduced deformability, the manner in which these properties interact with other factors—such
286 as crosslinking density or temperature—remains poorly understood. Future studies should

287 explore synergistic effects by investigating the manner in which the optimization of gel
288 concentration can enhance gel stability during industrial processing conditions.

289 3.3. pH

290 The effects of pH on the large deformation of gels are primarily mediated through pH's
291 effect on gel swelling, structural stability, and microstructure (Z. Liu et al., 2024). The response
292 of different gels to pH varies based on factors such as gel composition, crosslinking agents,
293 and environmental conditions. pH influences the structure of proteins by altering the balance
294 of attractive and repulsive forces, which can lead to protein unfolding and denaturation (Tan et
295 al., 2021). For example, Razi et al. (2019) studied egg white albumin and basil seed gum
296 mixtures under different pH conditions at high pressure. The gels formed at pH 5.0 exhibited
297 higher elasticity than those formed at pH 7.0, likely because of stronger electrostatic and
298 hydrophobic interactions at pH 5.0 (Hosseini-Parvar et al., 2016). These findings showed that
299 pH is a critical factor influencing the large deformation behavior of gels, highlighting its role
300 in shaping of gel properties.

301 Proteins, as natural polyampholytes, can form gels through heat-induced gelation or cold
302 gelation (Betz et al., 2012). The pH-dependent swelling behavior of protein gels has been
303 extensively studied. Saglam et al. (2013) observed that protein particles in gels formed via
304 emulsification and heat-induced gelation remained stable across a wide pH range but exhibited
305 aggregation at pH ~5 likely because of reduced electrostatic repulsion near the protein's
306 isoelectric point. This aggregation can cause substantial volume changes, affecting the
307 rheological properties of the gel. Wang et al. (2021) studied whey protein gels containing
308 glucose and fructose fabricated via the Maillard reaction at different pH levels. They observed
309 that the elastic modulus of these gels depended on the strength of interactions between the
310 protein molecules and sugar chains. The gels reached their maximum elasticity at different
311 temperatures, and their elasticity decreased as the pH increased, demonstrating that pH

312 influences molecular interactions within gels.

313 *3.4. Temperature*

314 The treatment temperature of gels is crucial for determining their strength and rheological
315 properties. Meat, a common component of the diet for several populations worldwide, exhibits
316 gel-forming and viscoelastic properties that depend on the bonds stabilizing the gel network
317 (Tammattinna et al., 2007). Changes in treatment temperature can affect the strength, rheology,
318 and sensory attributes of meat gels. For example, Mad-Ali et al. (2018) observed marked
319 differences in gel strength and deformation among goat meat gels treated at different
320 temperatures. They noted that preheating the meat prior to heating it to 90°C could enhance the
321 gel's breaking force and deformation capacity, highlighting the importance of temperature in
322 controlling meat gel behavior.

323 Starches too are key components of processed foods, and the gelatinization process during
324 heating affects the rheological behavior of starch gels (Rocha et al., 2011). Albano et al. (2014)
325 studied gels made from Peruvian carrot starch at different temperatures (10°C, 30°C, 50°C, and
326 70°C) and observed that they exhibited shear-thinning behavior and high thixotropy. Notably,
327 the results demonstrated that the setting temperature substantially affected the consistency
328 index of the gels. Further, Sha et al. (2019) observed that higher extraction temperatures of
329 porcine gelatin reduced the gel strength, gelation point, and melting point, demonstrating the
330 manner in which temperature affects the gelling properties of gelatin.

331 While pH and temperature are known to affect swelling and gelation kinetics, their
332 interaction with mechanical stress during large deformation remains underexplored. For
333 instance, acidic conditions can increase elasticity in egg albumin gels; however, the manner in
334 which these conditions interact with thermal cycles to influence fracture behavior is still unclear.

335 *3.5. High-pressure processing*

336 HPP, a nonthermal sterilization method in which pressures ranges from 100 to 1000 MPa,

337 is applied to food products, and it prevents spoilage without affecting the sensory or
338 physicochemical characteristics of food (Fam et al., 2020; C. Ren et al., 2020). This treatment
339 also provides functional effects and is widely used in the food industry (Mokrushin et al., 2022).
340 HPP can lower the gelatinization temperature of starch, with higher pressures providing greater
341 reductions in the gelatinization temperature (Balakrishna & Farid, 2020). In addition, HPP
342 treatment can be employed to obtain starch gels with more stable crystallites, denser structures,
343 reduced retrogradation, and increased gel stability (Guo et al., 2022).

344 Food products often comprise proteins and polysaccharides, which affect the texture,
345 structure, and stability of gels because of their gelling, thickening, and surfactant properties. In
346 mixed gels, HPP disrupts the interactions between proteins and polysaccharides, causing
347 biopolymer complexes to dissociate and proteins to denature, which alters gel properties (Razi
348 et al., 2019). Previously, HPP has been successfully applied in fish processing (Qin et al., 2022).
349 Moreover, this technique is employed to modify protein properties in several foods, altering
350 protein structure and affecting coagulation, aggregation, and gelation behavior (Moreno et al.,
351 2015).

352 Velazquez et al. (2021) showed that HPP can create a soft flexible network in crabmeat
353 gels, reducing their rigidity by fully denaturing the proteins and facilitating aggregation through
354 disulfide bonding. In particular, HPP at 100 MPa afforded the crabmeat gels with optimal
355 rheological properties. Similarly, Xu et al. (2024) observed that HPP caused the gradual
356 unfolding of α -helices in Tai Lake whitebait myofibrillar protein gels, transforming the α -
357 helices into β -sheets. Compared with untreated gels, these gels exhibited a 4.8-fold increase in
358 gel strength and improved elastic and viscous moduli (G' and G'').

359 3.6. Crystals

360 The large deformation properties of gels are influenced by natural crystallization during
361 aging and can also be affected by the addition of external crystals. Aging causes retrogradation

362 in gels, gradually altering their physical properties (Jiang et al., 2020). This process involves
363 the crystallization of molecules in the gel, which results in the formation of an ordered structure.
364 Retrogradation affects not only the texture, rheological behavior, and pasting properties of gels
365 but also their crystalline and molecular structures (X. Liu et al., 2021).

366 Inulin, an oligosaccharide, forms gels containing a network of small crystals when mixed
367 with water at high concentrations (J. Xu & Kenar, 2024). With time, these gels develop larger
368 crystals and their fracture strain decreases. The retrogradation temperature also plays a role in
369 this process: lower temperatures make the gels more brittle, and the maximum fracture force is
370 generated during large deformation. Starch gel retrogradation is influenced by the water content,
371 with higher water levels facilitating the formation of softer gels with well-defined crystallites
372 (X. Liu et al., 2021).

373 The addition of crystals also affects gel deformation. Xiao et al. (2020) added cellulose
374 nanocrystals (CNCs) to whey protein gels, increasing their water-holding capacity, strength,
375 viscoelasticity, and thermal stability. CNCs restricted the water mobility within these gels and
376 transformed the α -helixes of proteins into β -sheets. These findings were similar to those
377 reported by X. Liu et al. (2021) who highlighted the role of moisture in gel deformation. Thus,
378 naturally sourced CNCs are being explored as promising gel modifiers in the food industry.

379 *3.7. Crosslinking agents*

380 The water absorption capacity of a hydrogel depends on factors such as the type and
381 density of the polymer and crosslinking agent used (Juan et al., 2022). Crosslinking agents are
382 chemicals that link unsaturated compounds at the molecular level, forming a 3D network
383 structure in gels. These agents are used in various fields, including the food industry,
384 biomedicine, and materials science. In the food industry, enzymes and certain chemicals are
385 the most commonly employed crosslinking agents (Table 1). Enzymatic crosslinking agents
386 such as transglutaminase (TG) and various oxidases catalyze the formation of covalent bonds

387 among proteins (Zheng et al., 2022). Chemical crosslinkers are typically small molecules that
388 interact with proteins or other food components to yield a stable crosslinked framework, thus
389 influencing food characteristics, particularly food texture. Other than these conventional
390 crosslinkers, some specialized crosslinking agents can also be employed for specific
391 applications. For instance, a novel crosslinking agent that contains whey protein, papain,
392 glycerol, and epicatechin gallate exemplifies the recent advancements in crosslinking strategies
393 (Liu et al., 2023).

394 Gelatin, a biopolymer derived from collagen, is widely used in food and pharmaceuticals
395 owing to its swelling capacity, biodegradability, and biocompatibility (Echave et al., 2019).
396 Although gelatin can form hydrogels upon cooling, its mechanical properties are often poor,
397 necessitating the use of crosslinking agents (Uranga et al., 2020). Lin et al. (2023) observed
398 that TG can crosslink fish gelatin (FG) in a dose-dependent manner, notably increasing gel
399 strength (G'). Thus, the introduction of covalent crosslinks via TG can enhance the
400 thermodynamic and mechanical stabilities of FG gels, creating more organized structures.

401 In summary, crosslinking agents greatly improve the strength, stiffness, and stability of
402 gelatin hydrogels, which makes them valuable across various industrial sectors. However, the
403 influence of crosslinking agents on the large deformation of gels warrants further exploration
404 as this could help in the development of gels with improved properties and provide insights
405 into the internal factors governing their large deformation.

406 **Table 1.** Crosslinking agents used in common gels (Alavarse et al., 2022).

Crosslinker	Source	Polymers	Mechanism	pH, Temperature
Transglutaminase	Bacteria, mammals, and plants	Gelatin, collagen, whey protein, soy protein, chitosan, and carboxymethyl cellulose	Presence of glutamic acid and lysine groups; acyl ligands/amine ligands from L-glutamic acid and L-lysine	5.5, 40°C
Horseradish peroxidase	Root (<i>Armoracia rusticana</i>)	Hyaluronic-Tyramine (@Corgel), alginate–tyramine, gelatin norbornene/4-arm poly(ethylene glycol) dihydrogen tetrazine, and silk-fibroin	Oxidation of phenolic and catechol groups (requires hydrogen peroxide addition), peroxidation, isomerization, dimerization, and enolization	7.0, 45°C
Tyrosinase	Microorganisms, plants, and animals	Alginate, hyaluronic acid/gelatin, and glycol chitosan	Presence of phenolic/catechol groups. Oxidation (O ₂ species), semi-quinone formation through Micheal addition, isomerization, dimerization, and enolization	6.5, 50°C
Whey protein	Animals	Protein and polysaccharides	Maillard reaction	7.0, 50°C
Epicatechin gallate	Plants	Gelatin	Aldehyde group in gallic acid reacts with the amino group to form a stable covalent bond	5.5, 45°C

407 Although considerable efforts have been made toward understanding the specific

408 mechanisms through which these substances impact the large deformation in gels, continued
409 research is still needed. Some researchers are now employing computational mechanics to
410 examine the large deformation of gels, providing theoretical information that could aid the
411 understanding of and improvement in gel properties.

412 **4. Applications of the large deformation of gels**

413 Gels, as abundant renewable biopolymers, have attracted attention because of their
414 sustainability, affordability, and biocompatibility. Consequently, they have emerged as useful,
415 cost-effective, and easily modifiable materials (Li et al., 2021). Their unique properties and
416 wide application prospects have sparked interest across various sectors, including food (De
417 Leon Rodriguez & Hemar, 2020), biomedicine (Z. Li et al., 2020), soft robotics (Jiang et al.,
418 2023), and flexible electronics (Zhu et al., 2024). Gels have widespread applications in food
419 products (Fig. 3), and gel texture notably impacts the taste of foods. Exploring the large
420 deformation characteristics of gels, such as gel strength and viscoelasticity, can help in the
421 development of foods customized to consumer preferences. Moreover, gels can be used for the
422 production of easy-to-consume foods for individuals with chewing difficulties, thus addressing
423 their specific needs.

424 *4.1. Food texture*

425 Texture refers to the sensations experienced when eating food, including the initial bite,
426 chewing, and swallowing (Kutlu et al., 2022). Commonly referred to as “mouthfeel,” texture
427 plays a key role in consumer perception because it represents the sensory experience of a food’s
428 molecular, microstructural, and macrostructural composition. Therefore, texture is evaluated
429 through visual assessment, touch, and sound (Cao & Mezzenga, 2020). Among these factors,
430 the tactile feedback provided by the mouth during eating is the most important. The first bite,
431 which breaks down the structure of the food, providing different sensory perceptions based on
432 the gel’s fracture stress and strain (Franks et al., 2020). Gels with higher deformation capacity

433 provide a softer initial bite but remain resilient, providing a satisfying chew. The large
434 deformation behavior of gels also impacts the manner in which the gel breaks down and
435 releases flavors, which is a crucial consideration while designing foods to suit consumer
436 preferences. For example, softer gels are preferred in desserts, while firmer, more elastic gels
437 are used in products such as gummies or meat analogs for a chewier texture.

438 The texture of solid and semisolid foods is attributed to the mechanics of food breakdown
439 in the mouth (van Bommel et al., 2019). The consumer's first bite and chew are governed by
440 the large deformation properties of the food (Gravelle & Marangoni, 2021). Instruments can
441 be employed to analyze the fracture properties of gels, such as fracture stress and strain, which
442 are directly related to their texture. In particular, fracture stress is associated with the perceived
443 hardness of foods, while fracture strain is related to how easily the food deforms during the
444 first bite (Nishinari et al., 2024).

445 *4.2. 3D food printing*

446 3D food printing is an advanced process combining 3D modeling, electromechanical
447 control, food science, and other technologies to create food products through an additive
448 manufacturing process (Wang et al., 2022). This method enables the production of customized
449 foods that cater to specific consumer needs, enhancing their sensory qualities, nutritive value,
450 and texture (Park et al., 2020). Foods such as chocolates, cakes, candies, and even artificial
451 meat have been successfully produced via 3D printing (He et al., 2020). Compared with
452 traditional food production methods, 3D printing offers benefits such as lower cost, simpler
453 operation, and faster production of customized products (Umeda et al., 2024). During printing,
454 gels undergo notable shear deformation because they are extruded through the printer nozzle.
455 The large deformation properties of gels, including their yield stress and viscoelasticity, impact
456 their ability to maintain and recover their shape after deposition for 3D printing. Gels that
457 cannot recover well may collapse or lose their structure, resulting in printed products of poor

458 quality. Therefore, gels need to be able to withstand large deformations while retaining their
459 shape so as to ensure the successful 3D printing of complex food designs. [Kim et al. \(2022\)](#)
460 developed a low-calorie form of surimi by combining carrageenan, a highly elastic protein
461 substitute, with surimi through coaxial 3D printing. This process maintained the freshness and
462 tenderness of surimi while reducing its caloric content, making it ideal for older adults or
463 fitness-conscious individuals.

464 In recent years, 3D food printing has attracted significant attention, because of which
465 natural food gels are being developed for printing. The deformability of these gels allows for
466 us to modify their fluidity, elasticity, and thermal properties through additives or treatments,
467 enabling the fabrication of suitable materials for 3D printing. This technology can improve the
468 nutritional value of meals, benefiting both consumers without major health issues and those
469 with nutritional disorders such as malnutrition ([Anukiruthika et al., 2020](#)). Notably, 3D printers
470 can now create foods customized to consumer preferences by optimizing their flavor, texture,
471 color, price, and nutritional content. Natural substances, such as plant oils and emulsions, can
472 be used to modify the texture of printed food, influencing its printing behavior and gel
473 characteristics ([Yang et al., 2024](#); [Liang et al., 2024](#)). Developing natural modifiers for 3D
474 printing can enable the development of customized food products that promote health and cater
475 to a wide range of consumers, from children to older adults ([Das et al., 2023](#)).

476 Gels require high elasticity and stress recovery to maintain structural integrity during
477 extrusion. While shear-thinning additives provide improved printability, achieving consistent
478 postdeposition stability remains challenging. Future research should focus on hybrid gels that
479 offer both high printability and customized textural properties.

480 *4.3. Food preparation for individuals with dysphagia*

481 Dysphagia is defined as the difficulty or inability to swallow liquids, solids, or semisolids,
482 including drugs ([Nishinari et al., 2023](#)). This condition can lead to weight loss, malnutrition,

483 dehydration, and serious health issues such as aspiration pneumonia (Icht et al., 2018).
484 Individuals with dysphagia are advised to eat foods that are soft, moist, elastic, smooth, and
485 easy to swallow. Because gels are smooth, soft, and easy to break down, they are ideal for this
486 group of patients (Raheem et al., 2021). Gels prepared for patients with dysphagia should have
487 low fracture stress and should easily deform when chewed. Optimal large deformation
488 properties, such as high strain and low hardness, are important for controlling the texture and
489 consistency of these gels, making them easy to swallow without any risk of choking or
490 aspiration. By adjusting the viscoelastic properties of food gels via additives, these gels can be
491 customized to meet the needs of patients with different levels of dysphagia, ensuring safe
492 consumption. Thickeners modify the rheological properties of food, slowing its movement
493 through the mouth and throat, giving muscles more time to react and reducing the risk of
494 aspiration (Štreimikytė et al., 2020). Thus, gels can be used as thickeners to create softer, more
495 cohesive foods that are easier to chew and swallow.

496 Pure et al. (2021) modified the texture of a high-protein gel to make it suitable for patients
497 with dysphagia. After microwave processing, the gel became a soft solid. Fei et al. (2024)
498 developed a salmon-protein-based composite emulsion gel with konjac glucomannan and an
499 emulsion filler. Konjac glucomannan altered the protein structure, and the filler improved the
500 gel's elasticity and strength (Luo et al., 2024). These changes made the gel safer to swallow
501 and improved its cohesion, enhanced its mouthfeel, and reduced its stickiness.

502 Soft gels with controllable deformation properties are essential for individuals with
503 swallowing difficulties. The current formulations focus on texture but often neglect sensory
504 aspects such as flavor release and bioavailability. Thus, studies integrating rheological testing
505 with sensory analysis are needed to optimize these products.

506 4.4. Bionic foods

507 Gel-based bionic foods are an innovation in the food gel technology and can replicate the

508 structure and characteristics of natural foods (Zhou et al., 2023). They typically have a soft
509 texture and rich taste, and they mimic the flavor and texture of meats and other foods (Du et
510 al., 2023b). Plant-based meat alternatives are an important step in improving human health,
511 conserving resources, and supporting animal welfare (Sun et al., 2021). These alternatives are
512 also rich in essential nutrients, making them a great choice for daily nutrition. Moreover, the
513 development of sustainable bionic foods, such as meat substitutes, is crucial to meeting the
514 growing dietary demands, particularly in light of resource scarcity and environmental concerns
515 (Yang et al., 2023). Bionic foods can provide viable alternatives to natural foods, helping
516 mitigate these challenges (Du et al., 2022).

517 The ability of gels to deform without breaking is a key to simulating the texture of meat.
518 Bionic foods require gels that stretch, compress, and recover in ways similar to natural tissues.
519 Gel-based bionic foods that replicate the fibrous structure and chewiness of meat can be
520 developed by controlling the deformation properties, such as elasticity and toughness. This
521 replication is important for consumers' acceptance of plant-based alternatives. For example,
522 Du et al. (2023a) used agarose/konjac glucomannan double-network hydrogels to imitate the
523 texture of beef tripe. A 1:4 (w/w) ratio of agarose to konjac glucomannan resulted in a gel with
524 a texture similar to that of cooked beef reticulum and rumen, enabling the production of
525 artificial beef tripe.

526 Consumer surveys revealed widespread interest in the comparison of appearance, taste,
527 and texture of bionic foods with traditional ones (Hoek et al., 2011). While substantial progress
528 has been made in replicating the appearance and taste of natural foods by using flavor
529 compounds and plant pigments, the real challenge lies in recreating the texture, particularly
530 that of foods with complex structures. Through mechanical stretching, He et al. (2022)
531 developed an anisotropic hydrogel that successfully imitated the texture and taste of meat.
532 Likewise, for the simulation of the mouthfeel of other foods, bionic foods need to be processed

533 to fully utilize the large deformation properties of gels. The replication of anisotropic textures
534 in meat analogs requires gels with directional deformation properties. While recent advances
535 in protein–polysaccharide composites show promise, scalability and environmental impact
536 remain critical challenges to solve.

537 *4.5. Food packaging*

538 Hydrogels are useful as food packaging materials, particularly for controlling the moisture
539 content, which significantly impacts the shelf life of food products (Batista et al., 2019). These
540 gels are referred to as colloidal gels because their molecules are evenly dispersed in water.
541 Their viscoelastic and structural properties are derived from polymer chains and their
542 interactions with the surrounding medium (Ahmed et al., 2013). Packaging gels need to have
543 sufficient strength to withstand large deformations caused by handling, transport, and
544 environmental changes such as those in humidity. Their flexibility enables them to form a tight
545 seal around different food products, and their ability to retain water helps in moisture regulation,
546 extending the food’s shelf life.

547 Hydrogels for packaging are made from natural polymers such as proteins, carbohydrates,
548 and lipids, which can be sourced from plants or animals (Du et al., 2021). The main natural
549 polymers used to prepare hydrogels for food packaging are presented in Table 2, along with
550 their key features and properties. The proportion of these natural polymers can be adjusted to
551 prepare hydrogels with the desired water retention capacity, flexibility, and mechanical
552 properties (Jildeh & Matouq, 2020). Notably, hydrogels with good malleability and swelling
553 properties are suitable as food packing materials (Alharaty & Ramaswamy, 2020). Moreover,
554 these hydrogels need to exhibit good elasticity, tensile strength, and toughness, all of which
555 represent their large deformation properties.

556

557

558 **Table 2.** Properties of natural polymer-based hydrogels used in food packaging applications
 559 (Leyva-Jiménez et al., 2023).

Polymer	Material	Origin	Solubility	Advantages	Disadvantages
Polysaccharides	Chitosan	Chitin	Polar solvents	Great mechanical properties; great antimicrobial property; gas/water barrier properties	Highly hydrophilic
	Starch	Plants	Water	Highly biocompatible with polymers	Highly hydrophilic, brittle
	Pectin	Fruits	Water	High biodegradability; biocompatibility with polymers	Unstable mechanical properties
	Gum	Resins	Water	Emulsifying properties	Unstable mechanical properties; poor gas/water barrier properties
	Alginate	Marine algae	Water, ethanol	Self-healing properties; formation of water vapor barrier	Instability against radiation or handling
	Bacterial cellulose	Bacteria	Water	Film forming versatility; great gas/wate barrier	Laborious process of isolation
Protein	Zein	Corn	Ethanol, acetone	Film forming properties	Brittleness
	Soy protein isolate	Soybean	Water, ethanol	Gas barrier properties; pH- and temperature-responsive films	Water resistance, rigidity, brittleness
	Whey	Milk	Water, ethanol	Great mechanical properties; reduced oxygen and carbon dioxide permeability	Moderate barrier to gas

Polymer	Material	Origin	Solubility	Advantages	Disadvantages
Protein	Casein	Milk	Water	Forms flexible and nonopaque films; gas and lipid barrier properties	Brittleness
	Wheat	Starch	Water	Moderate barrier to oxygen	Brittleness
	Gelatin	Collagen	Water	Great gelling capacity; pH- and temperature-responsive films	Highly hydrophilic
Lipids	Wax	Bees	Water by emulsion	Low water vapor barrier	Poor mechanical properties
	Acetoglycerides	Fatty acid	Water by emulsion	Stretch properties; reduced water vapor permeability	Low mechanical strength; low oxygen permeability
Another	Lignin	Wood	Water by emulsion	Antioxidant capacity; great mechanical properties	Highly hydrophilic, brittle
	Poly lactic acid	Corn biomass	Organic solvents	Great mechanical properties	Brittle and weak films

562 Despite their potential, the development of natural hydrogels for food packaging is still in
563 its nascent stages and several challenges remain to be addressed. This has limited their
564 widespread application in food packaging.

565 **5. Conclusions**

566 Gels are important in the field of food science. These materials have specific structural
567 and functional properties that enhance the texture, taste, and stability of associated food

568 products. A comprehensive understanding of the factors influencing the large deformation
569 properties of food gels, including gel concentration, pH, temperature, and additives, is crucial
570 to innovating and developing gel-based products. Theoretical models have undergone
571 continuous refinement and improvement ever since the discovery of the large deformation of
572 gels. This review highlights the different factors impacting the large deformation of gels and
573 also discusses the theories and mathematical models developed to understand and simulate this
574 process. Furthermore, it summarizes the application of the large deformation characteristics of
575 gels in the fields of bionics and food science.

576 Although the application of gels with large deformation characteristics has grown in the
577 food industry, several challenges continue to persist. The process of gel deformation is very
578 complex and is influenced by a multitude of factors. The deformation of food gels during the
579 process of eating cannot be fully simulated by mathematical models because these models
580 cannot capture the intricacies of the process. Food consumption does not only involve
581 mechanical actions such as chewing but also the enzymatic action of saliva and the electrical
582 signals that transmit sensory information to the brain. Currently, this multifaceted process
583 cannot be completely simulated through mathematical models or mechanical simulations.
584 These problems need to be addressed in future research. We believe future research should
585 focus on combining theoretical models with experimental data so as to more accurately predict
586 gel behavior and develop blended gels customized for specific applications, such as 3D printing
587 and the fabrication of dysphagia-friendly foods. Sustainable and scalable innovations, such as
588 bio-based crosslinkers and mixed additive systems, should also be explored.

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594 **CRedit authorship contribution statement**

595 Bing Hu: Conceptualization, Formal analysis, Investigation, Writing–review and editing

596 Yulong Zhang: Methodology, Investigation, Writing–original draft

597 Lingyu Han: Conceptualization, Methodology, Writing–review and editing

598 Yiguo Zhao: Formal analysis, Writing–review and editing

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601 Jinxin Yang: Methodology, Software

602 Yapeng Fang: Methodology

603 **Declaration of competing interest**

604 The authors declare that they have no conflicts of interest regarding this study.

605

606 **Figure captions**

607 **Fig. 1.** Rheological responses in viscoelastic solids.

608 **Fig. 2.** Hydrogel undergoing deformation in response to external stimuli, and changes from the
609 reference state X_K to the current state x_i .

610 Note: Image created using BioRender and published with permission

611 **Fig. 3.** Applications of food gels.

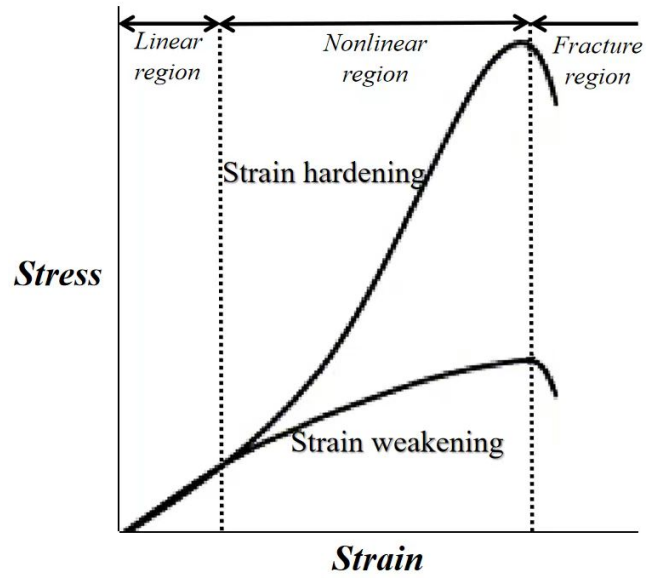
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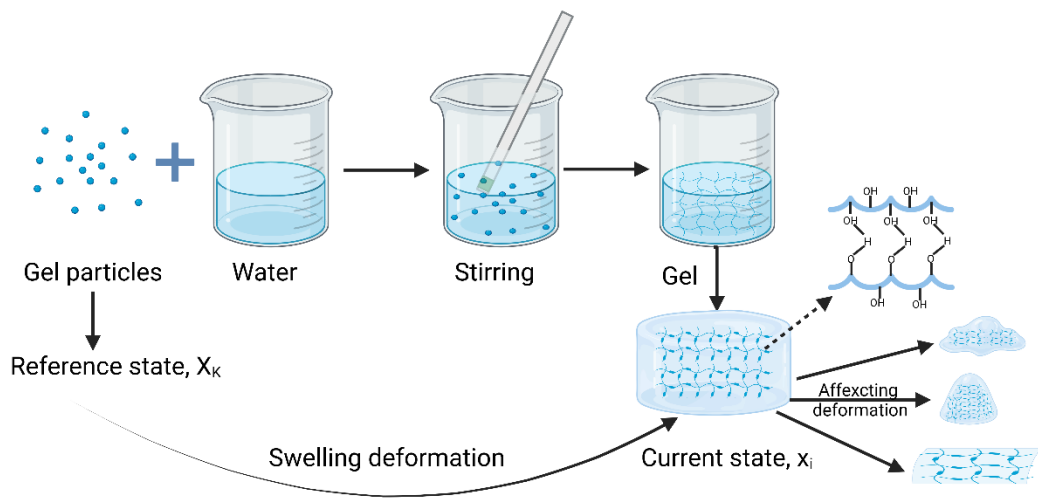
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Figures



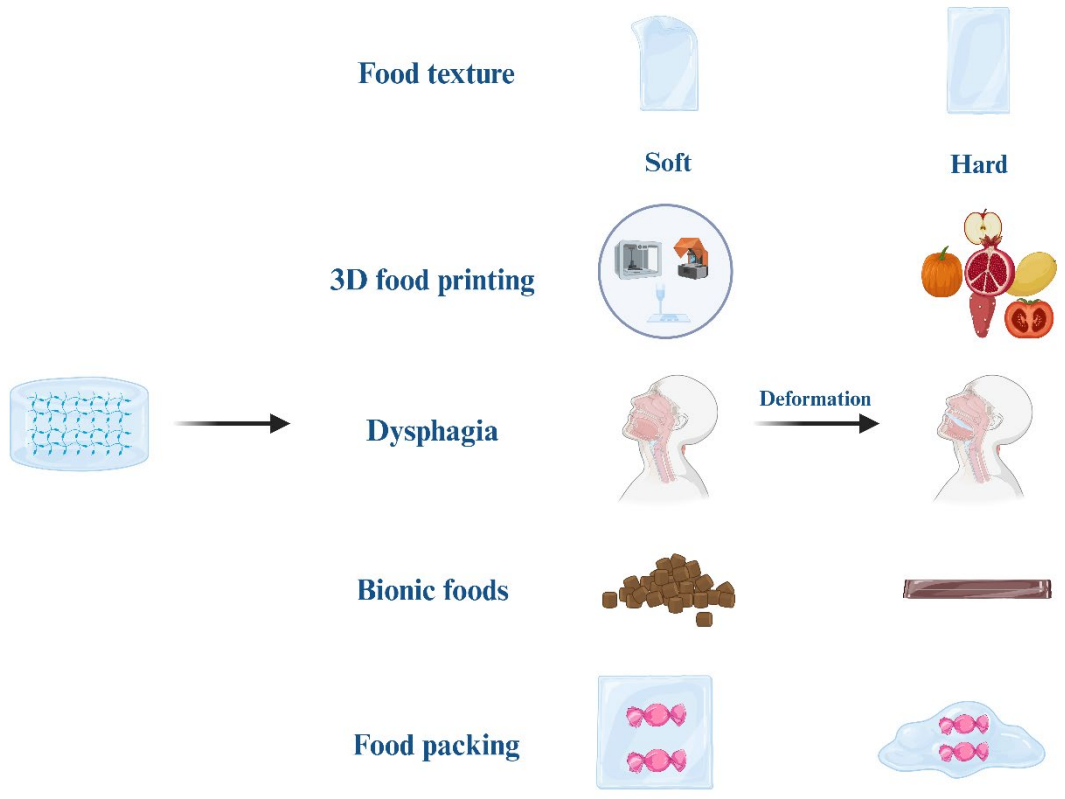
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Fig. 1



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Fig. 2



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Fig. 3

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